

2013 Ap Chem Answers

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AP[®] CHEMISTRY 2013 SCORING GUIDELINES Question 4 (15 points) For each of the following three reactions, write a balanced equation for the reaction in part (i) and answer the question about the reaction in part (ii). In part (i), coefficients should be in terms of lowest whole numbers.

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AP Chemistry 2013 Free-Response Questions

® AP CHEMISTRY 2013 SCORING GUIDELINES Question 1. (10 points) Answer the following questions about the solubility of some fluoride salts of alkaline earth metals. (a) A student prepares 100. mL of a saturated solution of MgF₂ by adding 0.50 g of solid MgF₂ to 100. mL of distilled water at 25°C and stirring until no more solid dissolves.

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calorimeter is calculated correctly and the final answer is correct (correct math, number of significant figures, and unit). Part (d) earned 2 points. The setup of $\Delta H^\circ = q/n$ is correct. Numbers are substituted into the equation correctly to get the answer with the correct number of significant figures and the negative sign.

AP CHEMISTRY 2013 SCORING GUIDELINES - College Board

AP[®] CHEMISTRY 2013 SCORING GUIDELINES ... Answer the following questions involving the stoichiometry and thermodynamics of reactions containing aluminum species. $2 \text{Al}_2\text{O}_3(\text{l}) + 3 \text{C}(\text{s}) \rightarrow 4 \text{Al}(\text{l}) + 3 \text{CO}_2(\text{g})$ An electrolytic cell produces 235 g of Al(l) according to the equation above.

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Use the information in the following table to answer parts (c) and (d). Name Lewis Electron-Dot Diagram Boiling Point (°C) Vapor Pressure at 20°C (mm Hg) Dichloromethane 39.6 353 Carbon tetrachloride 76.7 89 (c) Dichloromethane has a greater solubility in water than carbon tetrachloride has. Account for this observation

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Unfortunately, for AP Chemistry, most of the available official practice materials are for the old version of the test (pre-2013), but these can still be

useful for practice. You should be able to get newer practice tests from your teacher or through review books.

Every AP Chemistry Practice Test Available: Free and Official

The AP Chemistry Exam covers a full-year introductory college course in chemistry with laboratory. The multiple-choice questions in Section I cover the breadth of the curriculum. Section II, the free-response part of the exam, includes three quantitative problems, one question on writing chemical reactions and predicting products, and two essays.

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AP Chemistry 2013 Free Response Question 3

AP Chemistry Free Response Answers. Adrian has 30 years of high school and early college chemistry teaching experience, in both the UK and the USA.

AP Worked Answers - Adrian Dingle's Chemistry Pages

2013 ap' chemistry free-response questions chemistry section 11 (total time—95 minutes) part a time —55 minutes you may use your calculator for part a. clearly show the method used and the steps involved in arriving at your answers.

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answer. ii. After the new equilibrium is established at 575 K, is the value of K_p greater than, less than, or equal to the value before the $\text{NH}_3(\text{g})$ was injected into the ... 2013 AP Chemistry Free Response - Form B.docx. 2013 AP Chemistry Free Response - Form B.docx. Sign In Page 1 of 5 ...

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